THE SYNTHESIS OF DIASTEREOMERS OF PHOSPHOROTHIOATE ANALOGUE OF DIPALMITOYLPHOSPHATIDYLINOSITOL

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Abstract: The synthesis of diastereomers of 1,2-dipalmitoyl-sn-glycero-3-thiophospho-1'--inositol (DPPsi) has been accomplished via the phosphitylation and chromatographic separation of phosphorothioate triesters.

Chemical synthesis of inositol phospholipids and of myo-inositol phosphates has been the subject of considerable interest, due to the recent results on the role of inositol phospholipids and phosphates in cellular signalling pathways¹.

Numerous methods of the synthesis of the key representatives of inositol phospholipids diacylphosphatidylinositol, and of its analogues have been lately described², including the phosphorothioate analogue (DPPsI)³. In view of the documented stereospecificity of phospholipiases towards phosphorothioate analogues of phospholipids⁴, the diastereomers of DPPsI may prove useful antimetabolites blocking the receptor-mediated inositol phosphate metabolism. We now describe the method for chemical synthesis of individual diastereomers of DPPsI.

Thus, 1,2-dipalmitoyl-sn-glycerol 1 was undergone the reaction with chloro-N,Ndiisopropylaminomethoxyphosphine 2 in the presence of triethylamine in CH₂Cl₂ as described recently⁵. The resulting phosphoramidite was further condensed with D-(-)-2,3,4,5,6-pentabenzyl-myo-inositol⁶ (3) in THF-acetonitrile in the presence of tetrazole⁸. The product was sulfurized with elemental sulfur and the mixture of phosphorothioates 4 was deprotected using sequentially trimethylamine (to demethylate phosphorothioate function) and BF₃-ether in ethyl mercaptane⁹ (to deprotect inositol hydroxyl functions). Finally, pure DPPsI (5) was obtained by chromatography on silica gel as the mixture of diastereomers.



The attempts at separating this mixture by means of chromatographic techniques were unsuccesful. The total yield of isolated 5 was 47% (with respect to 3). Using the above scheme we have carried out the synthesis of natural DPPI with 61% total yield¹⁰. In this case, benzyl protective groups were removed by hydrogenolysis.

The separation of diastereomers 5 was finally achieved at the stage of the fully protected triesters 4 using column chromatography on silica gel and CCl₄-acetone (100:1) as eluting solvent (R_F: 4a 0.48, 4b 0.43, CCl₄-acetone, 40:1¹¹). Separated diastereomers 4a,b were deprotected as described above yielding pure 5a (δ 57.45 ppm) and 5b (δ 57.05 ppm), respectively. Both diastereomers were fully characterized by their ¹H NMR spectra¹² and by their hydrolysis reactions catalyzed by phospholipase A₂ (PLA₂) and C. ¹H NMR spectra of 5a and 5b¹² and of DPPI¹⁰ indicated that these compounds possess virtually identical conformation in the unaggregated state. The hydrolysis of 5a in the presence of

PLA₂ from bee venom in mixed micellar system (Figure 1) gave rise to palmitic acid and lysophospholipid 6 as opposed to 5b for which very little of such products could be detected by TLC.



Figure 1. ³¹P NMR spectra of the reaction mixture of (Sp + Rp)-DPPsI (10 mg) with PLA₂; other conditions : 50 mM HEPES.Na buffer, pH 7.2, 0.25 mM EDTA, 2 mM CaCl₂, 4% Triton X-100, 20% D₂O, temperature 310 K; each spectrum was acquired during 10 min.; curve e is an expansion of the signal in the spectrum d.

The time course of the enzymatic reaction is shown in Figure 1. The decrease in the intensity of the low field signal (5a) is accompanied by the increase of the other resonance due to coincidence of the chemical shifts of the signal from the undigested substrate 5b and the lyso-compound 6 (spectrum 1e). In accordance with the previously reported results, phospholipase C from Bacillus cereus catalyzed the hydrolysis of 5b, while no reaction was obtained with 5a. Therefore, based on the previously determined enzyme stereospecificities⁴, the configuration Rp is assigned to 5b and Sp to 5a.

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References and Notes

- 1. M.J.Berridge, *Biochem.J.*, 220, 345 (1984); B.Michell, *Nature*, 319, 176 (1986); M.J.Berridge and R.F.Irvine, *Nature*, 312, 315, 1984
- V.I.Shvets, A.E.Stepanov, V.N.Krylova, and P.V.Gulac, "Mio-Inozit i Fosfoinozitidy", Nauka, Moscow 1987, D.C.Billington, Chem.Soc.Rev., 18, 83 (1989)
- 3. G. Lin, and M.-D.Tsai, J.Am.Chem.Soc., 111, 3099 (1989)
- 4. K.S.Bruzik, R.-T.Jiang and M.-D.Tsai, Biochemistry, 22, 2478 (1983)
- 5. K.S.Bruzik, G.M.Salamonczyk and W.J.Stec, J.Org.Chem., 51, 2368 (1986)
- 6. 3 {[α]_D²⁰ -13.1°, (c3.8, CHCl₃)} was obtained from D-3,4,5,6-tetrabenzyl-myo-inositol ⁷ as described by R.Gigg, and C.D.Warren, J.Chem.Soc. (C), 2367 (1969)
- 7. D-3,4,5,6-tetrabenzyl-myo-inositol was obtained as recently described, K.S.Bruzik, G.M.Salamonczyk, Carbohydr.Res., 000 (1989)
- 8. M.D.Mateucci, M.H.Caruthers, J.Am.Chem.Soc., 103, 3185 (1981)
- 9. K.Fuji, K.Ichikawa, M.Node, E.Fuijta, J.Org.Chem., 44, 1661 (1979)
- DPPI: {[a]_D²⁰ +9.3° (c0.3, CHCl₃), m.p. 181° (dec.), lit.[a]_D²⁰ +7.48°, B.A,Klyashchitskii, E.G.Zhelvakova, V.I.Shvets, R.P.Evstigneeva, and N.A.Preobrazhenskii, *Tetrahedron Lett.*, 587 (1970); ¹H NMR, 300 MHz, CD₃OD): δ 5.26 (m, H-2), 4.45 (dd, H-1, J₁₂ 3.3 Hz), 4.208 (t, H-2', J_{1'2} = J_{2'3} 2.8 Hz), 4.213 (dd, H-1, J₁₂ 6.7 Hz), 4.079 (m, H-3), 3.913 (ddd, H-1', J_{1'2} 2.7 Hz, J_{1'P} 8.2 Hz, J_{1'6} 9.6 Hz), 3.796 (t, H-6', J_{6'}, = J_{6'5} 9.7 Hz, J_{6'P} 1 Hz), 3.656 (t, H-4', J_{3'4'} = J_{4'5} 9.5 Hz), 3.397 (dd, H-3'), 3.225 (t, H-5'), 2.344 (t, H-2", J 7.4 Hz), 2.319 (t, H-2", J 7.5 Hz), 1.62 (m, H-3"), 1.29 (m, CH₂), 0.893 (t, CH₃)
- 11. 4a: δ 31P 71.03 ppm; 4b: δ 31P 70.5 ppm
- 12. ¹H NMR (CD₃OD); **5a**: δ 5.26 (dd, H-2, J₂₃ 5.0 Hz), 4.44 (dd, H-1, J₁₂ 3.3 Hz), 4.256 (t, H-2', J₁₂ + J₂₃ 2.8 Hz), 4.213 (dd, H-1, J₁₂ 6.9 Hz), 4.129 (m, H-3, J₂₃ 5.1 Hz), 3.80 (t, H-6', J₆₁' = J₆₅ 9.6 Hz, J_{6P} 1 Hz), 3.644 (t, H-4', J₃₄ = J₄₅ 9.5 Hz), 3.39 (dd, H-3'), 3.237 (t, H-5'), 2.345 (t, H-2'' J 7.4 Hz) 2.317 (t, H-2'', J 7.5 Hz), 1.62 (m, H-3''), 1.29 (m, CH₂), 0.893 (t, CH₃); 5b: δ 5.267 (ddt, H-2, J₂₃ 5.0 Hz), 4.445 (dd, H-1, J₁₂ 3.3 Hz), 4.274 (t, H-2', J₁₂ = J₂₃ 2.5 Hz, J₂p<1.0 H), 4.203 (dd, H-1, J₁₂ 6.7 Hz), 4.134 (m, H-3, J₂₃ 5.3 Hz J 7.7 Hz), 3.79 (t, H-6', J₆₁'' = J₆₅ 9.3 Hz, J_{6P} 1 Hz), 3.637 (t, H-4', J₃₄ = J₄₅ 9.4 Hz), 3.39 (dd, H-3'), 3.222 (t, H-5'), 2.342 (t, H-2'' J 7.4 Hz), 2.316 (t, H-2'', J 7.6 Hz), 1.62 (m, H-3''), 1.29 (m, CH₂), 0.893 (t, CH₃)
- 13. The change in relative priority of glyceryl substituent in DPPsC with respect to DPPsI results in the inverse configuration assignment of PLA2 hydrolyzable isomer of DPPsI as compared to DPPsC

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